

Photocatalytic Abatement of Biofilms and VOCs
for
Cultural Heritage
Museum Laboratories
Human Health and Safety

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Destructive Biological Growth...



Can be Cleaned...

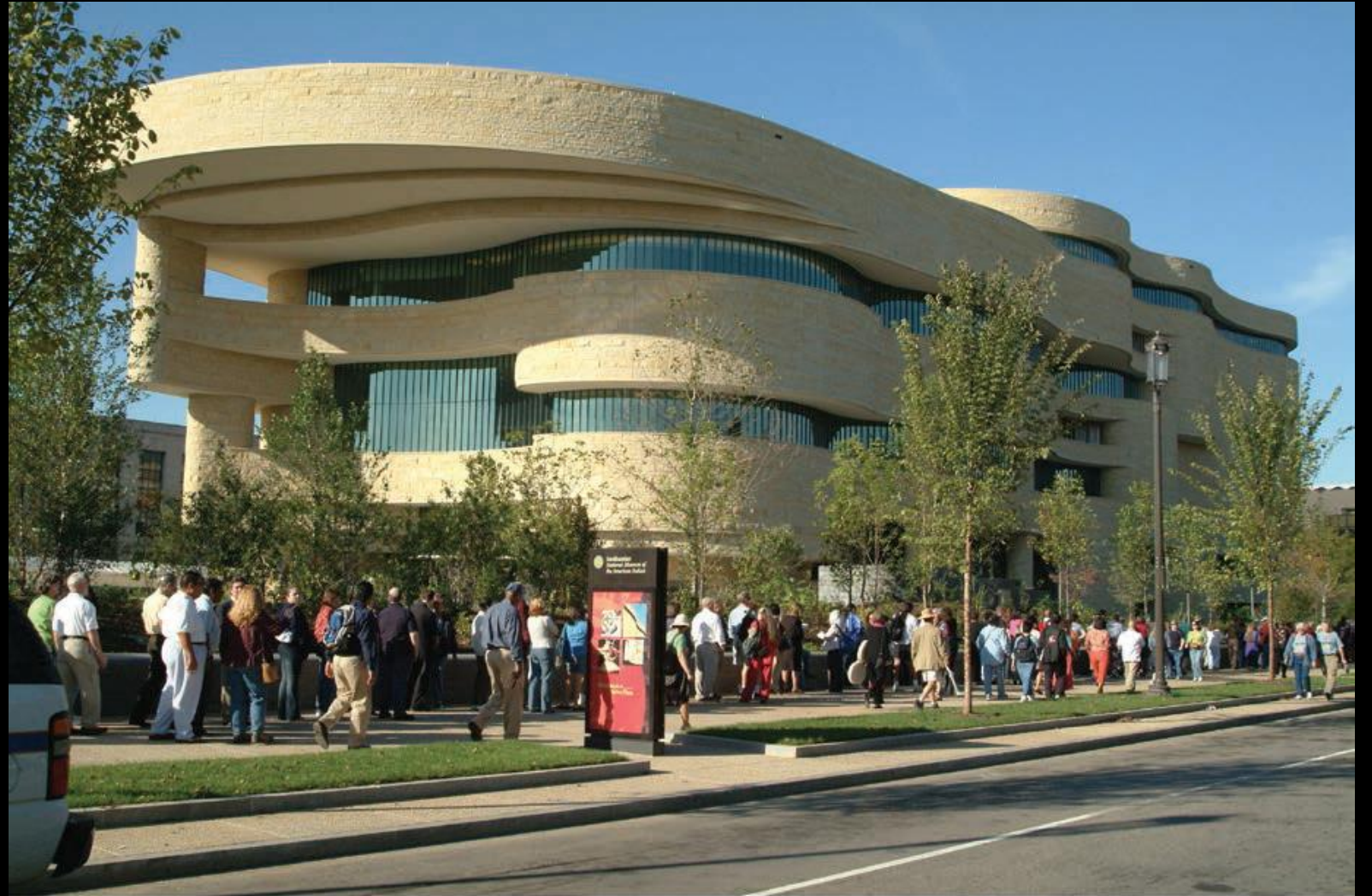


But Will Return

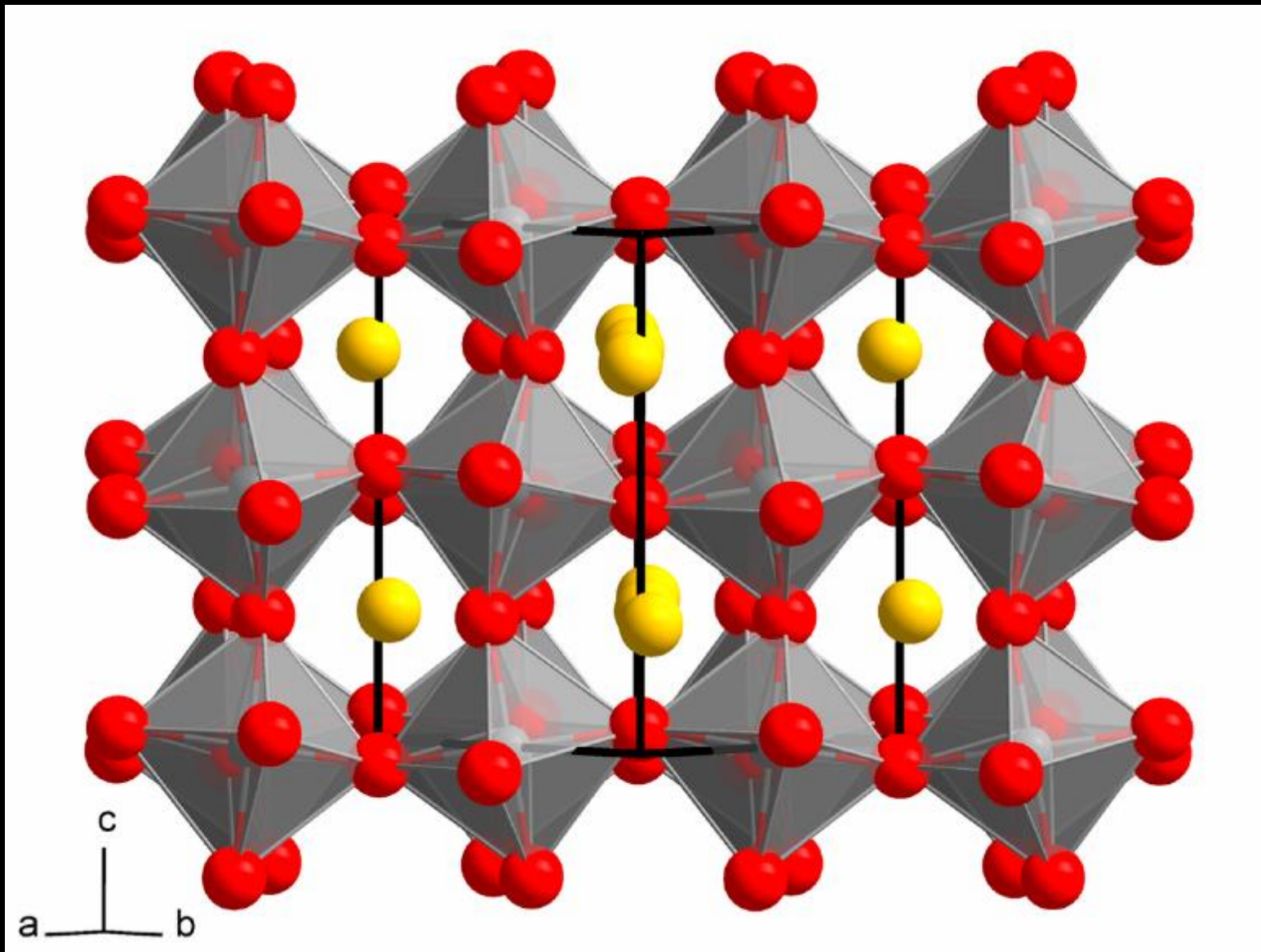
14 years old

Cleaned 3 times

Source:
Museum Conservation Institute
Smithsonian Institution
Photo by



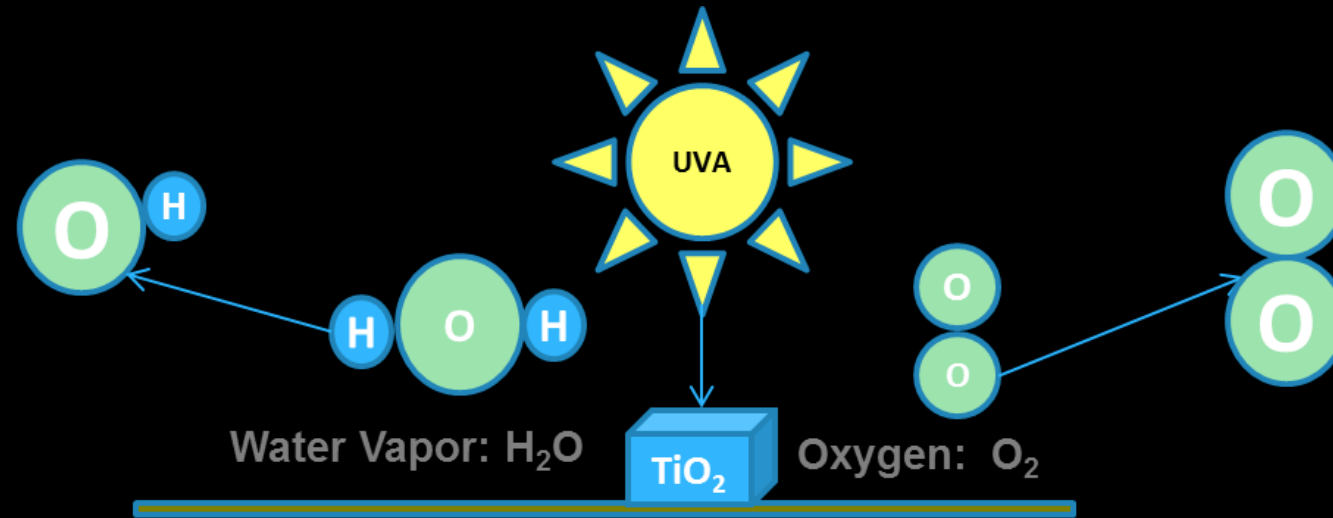
Anatase TiO₂



Effects of Photocatalysis

- Ultra Violet A creates “band gaps” on the crystal lattice.
- Breaks water (H_2O) into OH^\cdot (hydroxide radicals) and H^+ (Hydrogen ion)
- Transforms atmospheric Oxygen (O_2) into a Super Oxide Anion (O_2^-)

Photocatalytic Oxidation

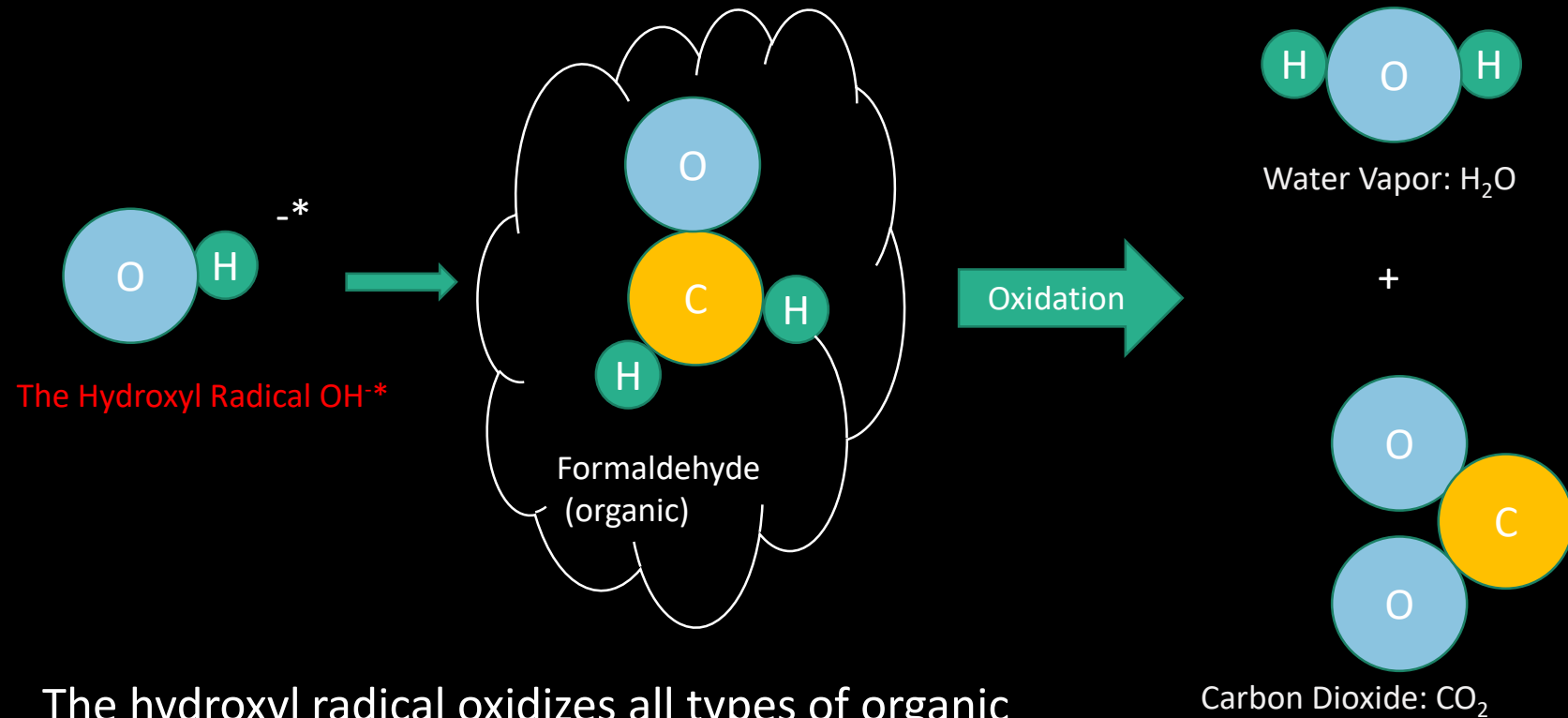


The Hydroxyl Radical OH^{*}

The hydroxyl radical is the most powerful, non-poisonous scrubbing agent in nature - stronger than straight 100% chlorine in oxidative power.

The Super Oxide Anion O₂^{*} (Activated Oxygen)

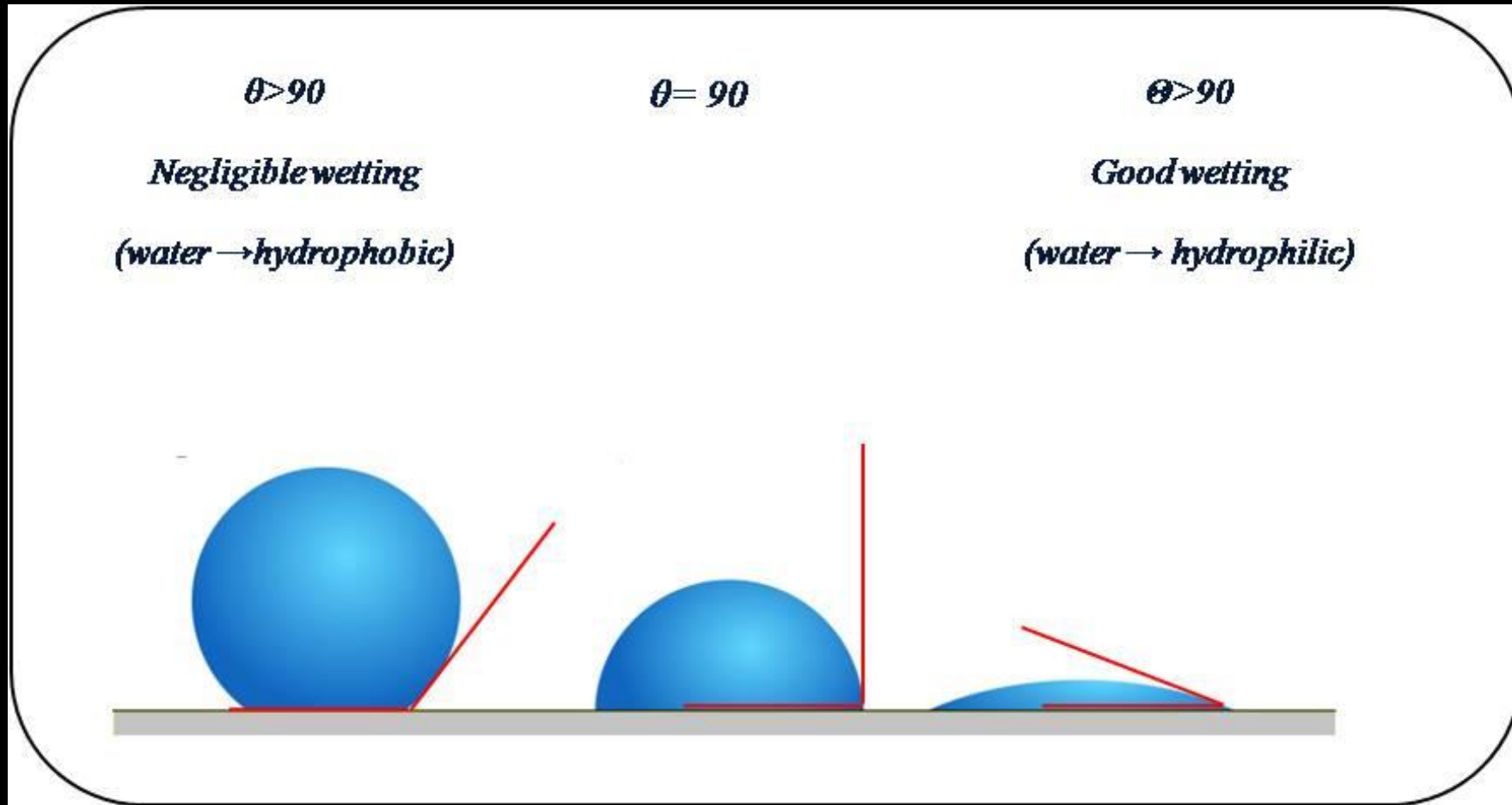
Photocatalysis: Oxidation Reaction



The hydroxyl radical oxidizes all types of organic compounds, including VOCs and formaldehyde, breaking them down to CO_2 and water vapor.

Water Contact Angle

Sheeting Action Leads to Self-Cleaning Exteriors





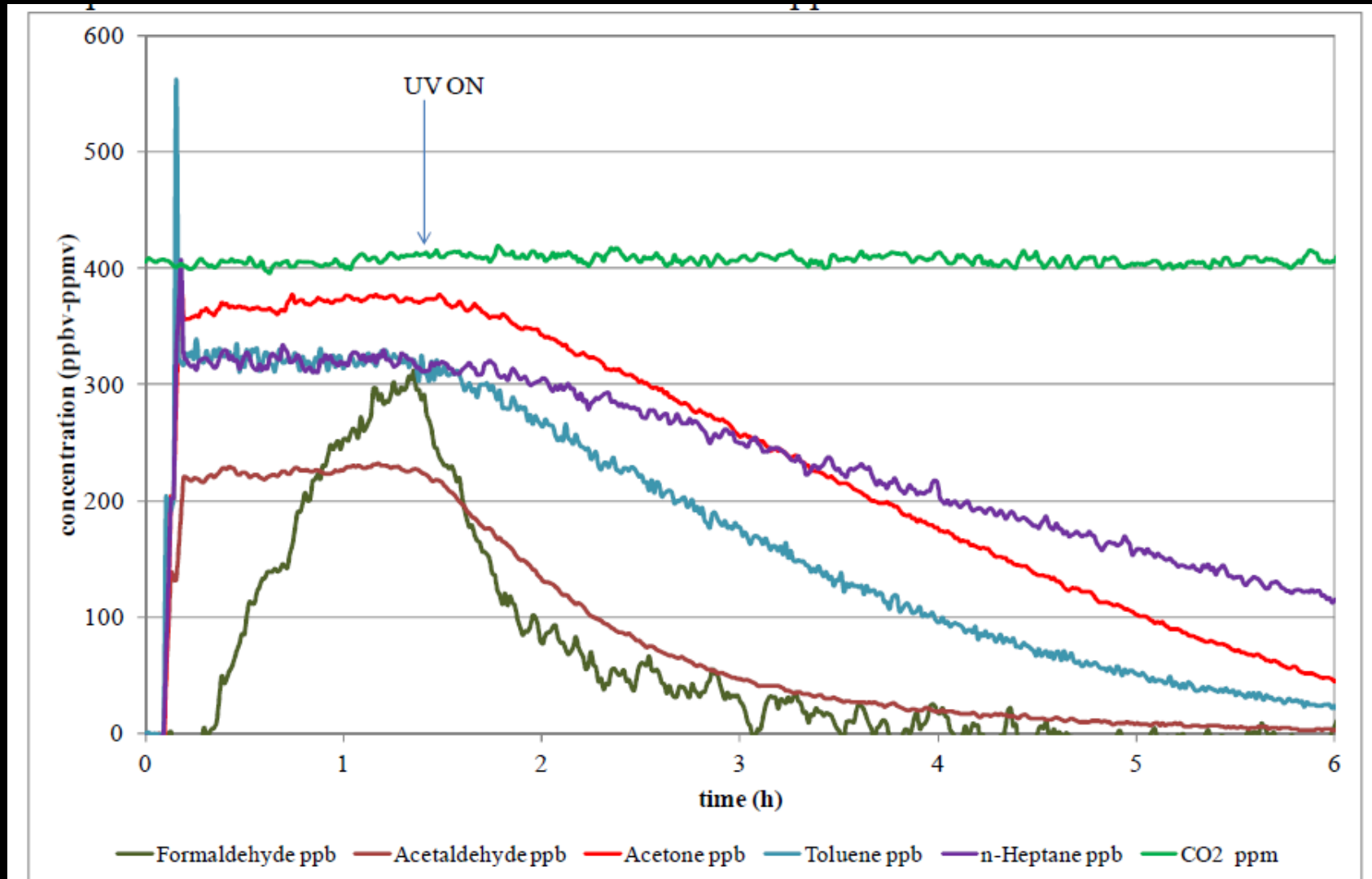
LOUISA C.
1851 - 1932

LILLIAN E.
1875 - 1963

Requirements

- Humidity
- UVA (natural or artificial)
- Circulation
- Catalyst
- Surface Area

Eliminates VOCs in the Presence of UV-A



Graph 2: VOC evolution versus time at low concentration.

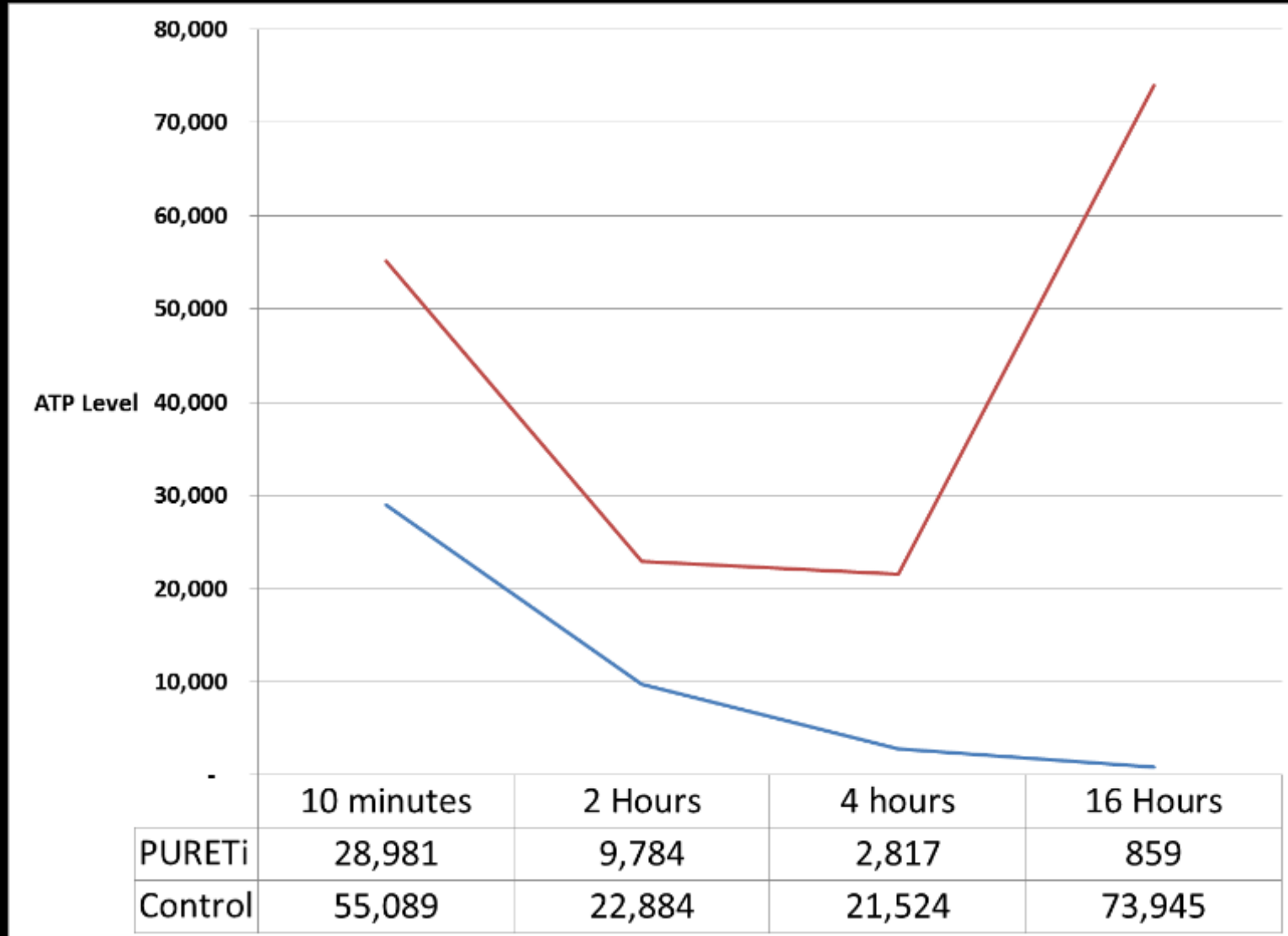




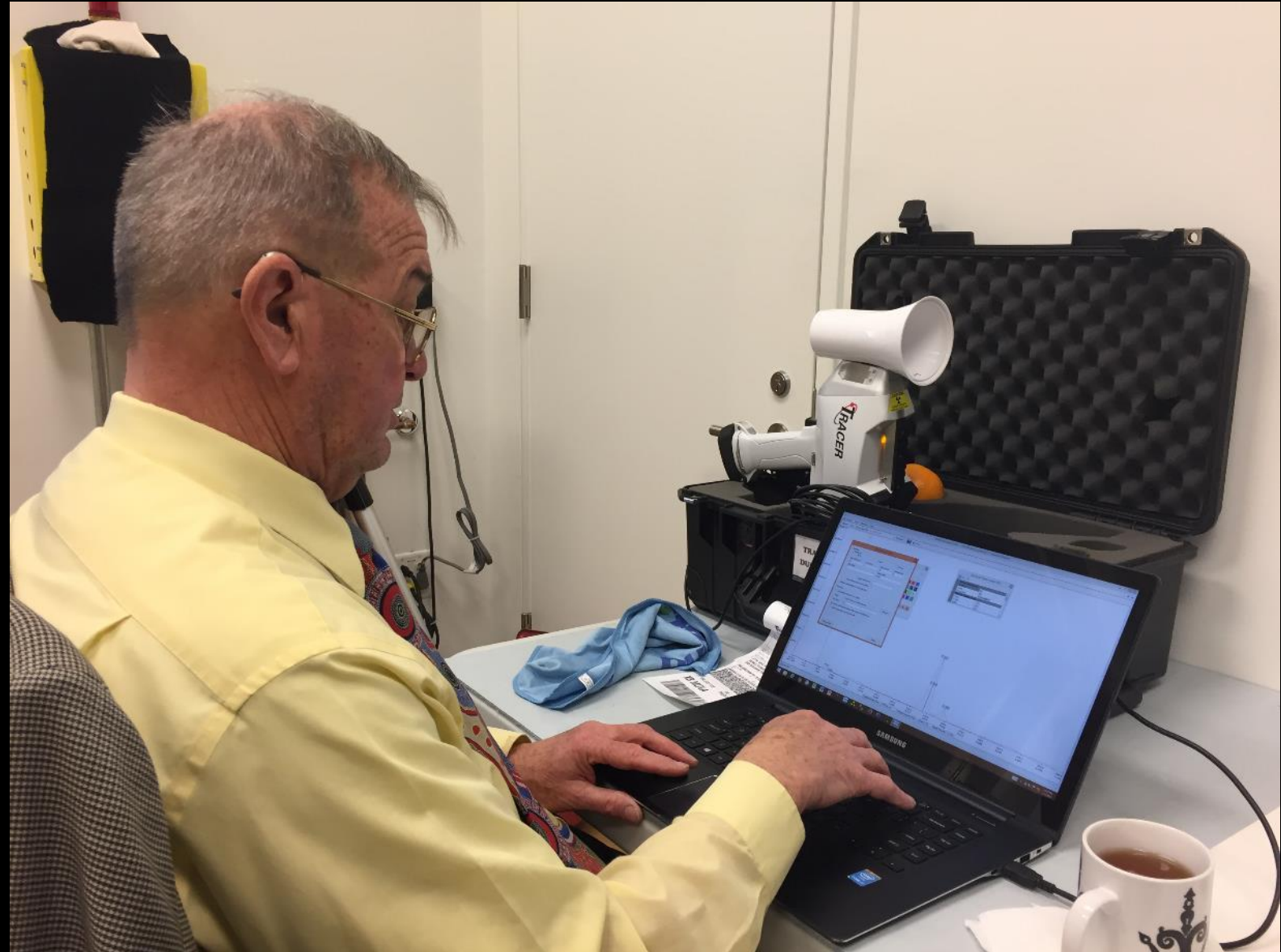




Reduction in Contact Surface ATP Levels

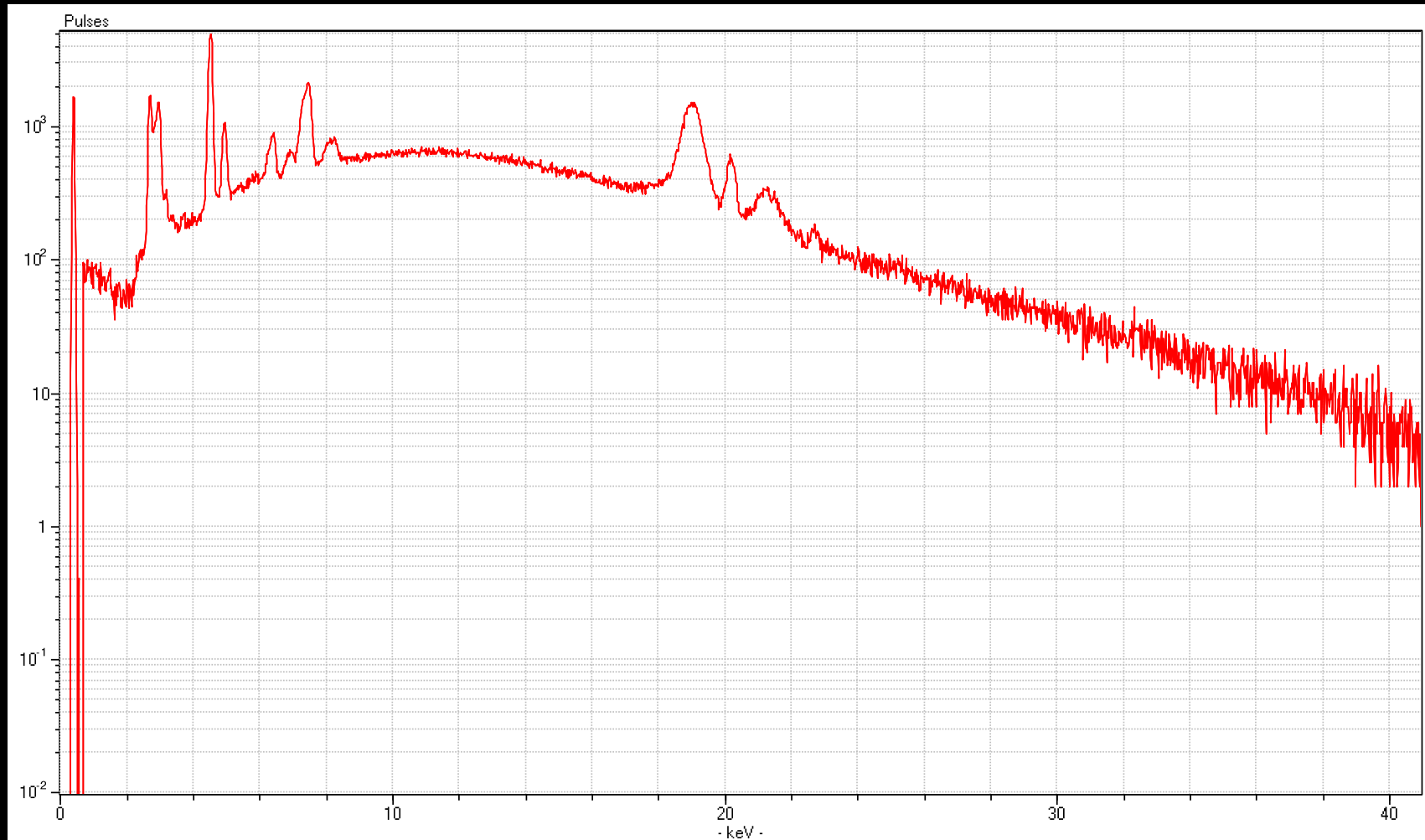


Validation using X-Ray Fluorescence





X-Ray Fluorescence Spectra







Source:
Museum Conservation Institute
Smithsonian Institution
Photo by Carol A. Grissom